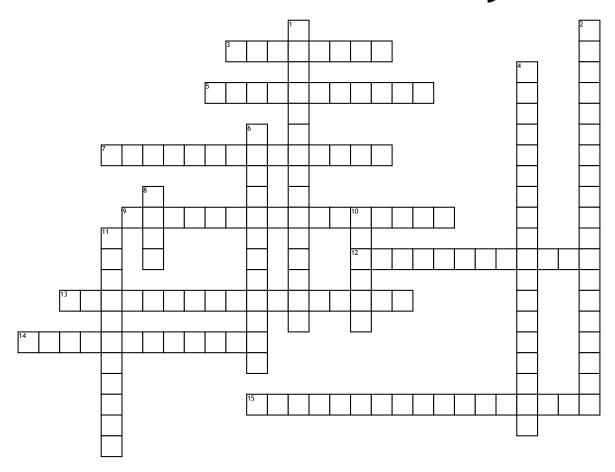
Name:	Date:	Period:

Thermochemistry



Across

- **3.** a system at constant pressure's heat content
- **5.** measurement for heat flowing in and out of a system for physical and chemical processes
- **7.** in a chemical equation it is the exact enthalpy change
- **9.** when completely burning a mole of a substance, it is the heat of the reaction taking place

- **12.** everything else that is not the focus
- **13.** gives off heat into the surrroundings
- **14.** the total heat required to increase the temperature of 1g of the substane 1 degree C
- **15.** the heat absorption of a solid melting at a constant rate

Down

1. changes in energy that happen in chemical reactions and in changes of state

- **2.** dissolution of a mole that cause enthalpy change
- **4.** takes in heat from the surroundings
- **6.** the total heat required in order to increase an object's temperature by 1 degree C
- **8.** energy transfer due to a difference in temperature
- 10. the focus of attention
- **11.** measuring device for the increase or decrease of heat in physical or chemical processes