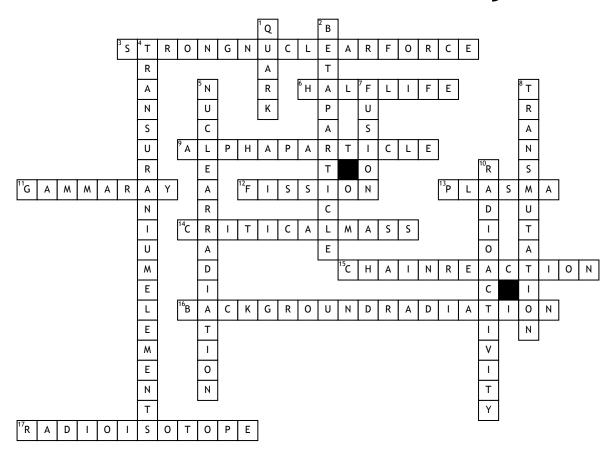
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nuclear chemistry



Across

- **3.** the attractive force that bonds protons and neutrons together in the nucleus
- **6.** the time required for 1/2 of a sample of a radioisotope to decay
- **9.** a positively charged particle made up of 2 protons and 2 neutrons
- 11. a penetrating ray of energy emitted by an unstable nucleus
- **12.** the splitting of an atomic nucleus into smaller parts
- **13.** a state of matter in which atoms have been stripped of their electron

- **14.** the smallest possible mass of a fissionable material that can sustain a chain reaction
- **15.** neutrons released during the splitting of an initial nucleus trigger a series of nuclear fission
- **16.** occurs naturally on the environment
- **17.** any atom containing an unstable nucleus

Down

- 1. a subatomic particle theorized to be among the basic units of matter
- **2.** an electron emitted by an unstable nucleus

- **4.** elements with atomic number greater than 92
- **5.** charged particles and energy that are emitted from the nuclei of radioisotope
- **7.** a process in which the nuclei of 2 atoms combine to form a larger nucleus
- **8.** the conversion of atoms of one element to atoms of another
- **10.** the process in which an unstable atomic nucleus emits charged particles and energy