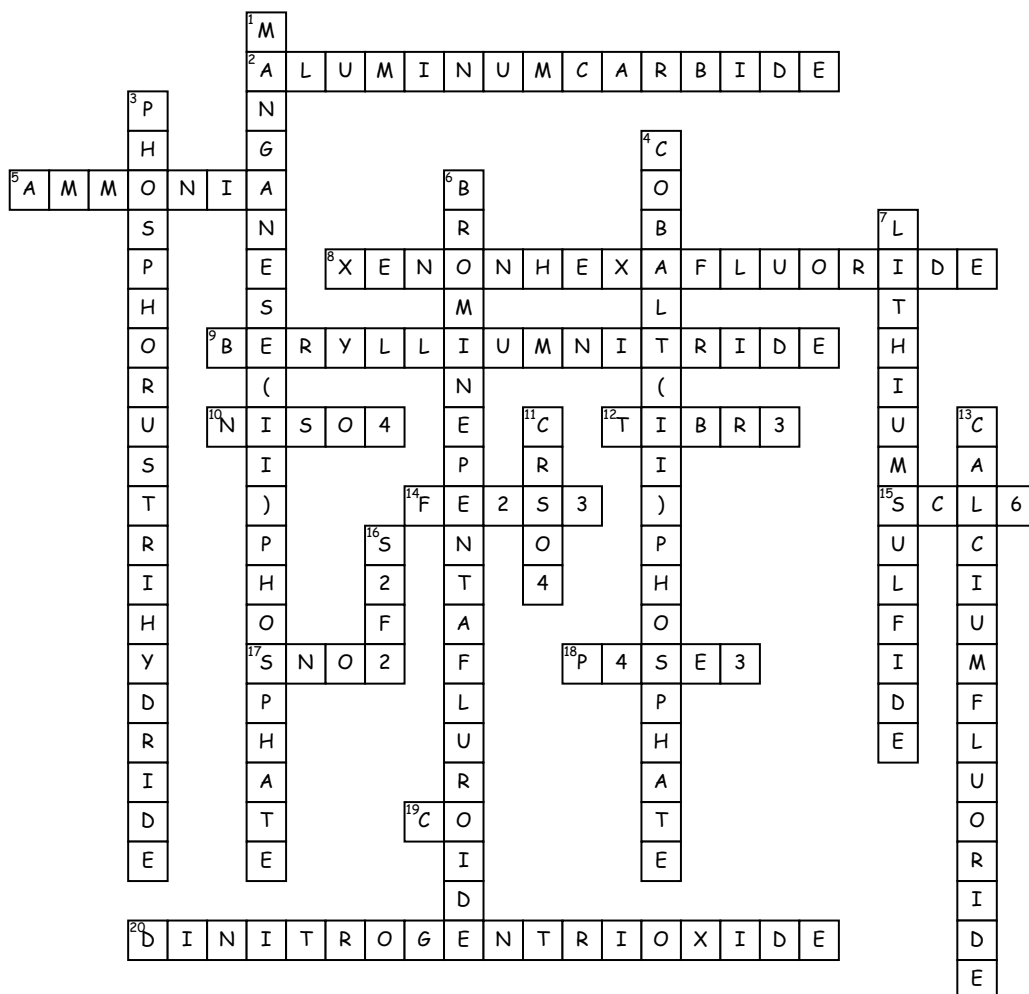


Name: \_\_\_\_\_

# Naming Ionic and Covalent Compounds



## Across

2.  $Al_4C_3$
5.  $NH_3$
8.  $XeF_6$
9.  $Be(NO_3)_2$
10. Nickel (ii) Sulfate
12. Titanium (iii) Bromide
14. Iron (iii) Sulfide

15. Sulfur Hexachloride

17. Tin (iv) Oxide

18. tetraphosphorus triselenide

19. Carbon Monoxide

20.  $N_2O_3$

## Down

1.  $Mn_3(PO_4)_2$

3.  $PH_3$

4.  $Co_3(PO_4)_2$

6.  $BrF_5$

7.  $Li_2S$

11. Chromium (ii) Sulfate

13.  $CaF_2$

16. Disulfur Difluoride

## Word Bank

Beryllium Nitride

$SnCl_6$

Manganese (ii) Phosphate

Ammonia

$Fe_2S_3$

$TiBr_3$

Cobalt (ii) Phosphate

Phosphorus Trihydride

Xenon Hexafluoride

$SnO_2$

$S_2F_2$

$CO$

Calcium Fluoride

Aluminum Carbide

Bromine Pentafluoride

$CrSO_4$

$NiSO_4$

$P_4Se_3$

Lithium Sulfide

Dinitrogen Trioxide