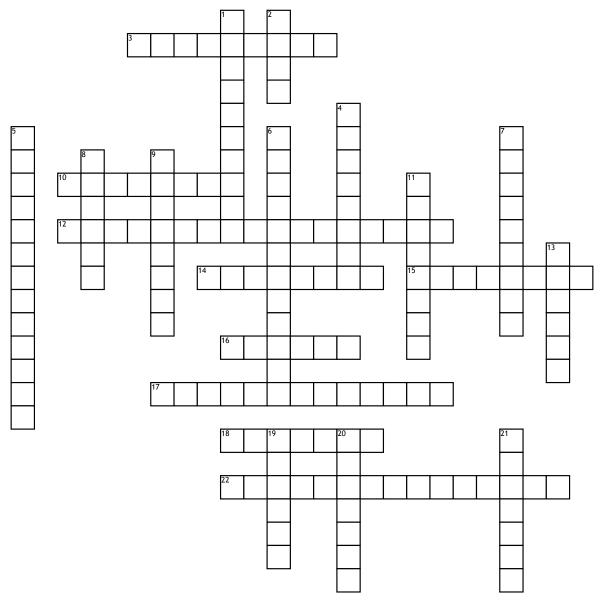
Name:	Date:
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Basic Chemistry



Across

- 3. The study of matter
- **10.** Substances that affect the rate of a reaction but are not changed themselves. Also may start or stop a reaction from occurring.
- **12.** Shows how atoms in a molecule are located and connected
- **14.** Substances that enable enzymes to work properly.
- **15.** Pure substances consisting of two or more atoms in each molecule
- **16.** Positive particle of an atom. The number of these determines what kind of element it is
- **17.** Chart of elements organized according to their atomic structure

- **18.** Pure substances consisting of only one kind of atom in each molecule.
- **22.** Show the number and type of atoms in a molecule

Down

- 1. Small amount of liquid chemicals would be measured using this metric unit.
- 2. The smallest unit of an element
- **4.** The metric unit used to weigh of a bar of gold
- 5. Solid, liquid, gas
- **6.** Shared electrons hold atoms together.

- **7.** Donated electrons form charged particles called ions to stick together
- **8.** Occupies space and has mass
- **9.** Negatively charged particle outside the nucleus
- **11.** A distinct group of atoms bonded together
- **13.** A protein molecule or organic molecule used as a catalyst
- **19.** The ability to do work
- **20.** The part of the atom containing the protons and neutrons
- 21. Neutrally charged particle found in the nucleus and has mass