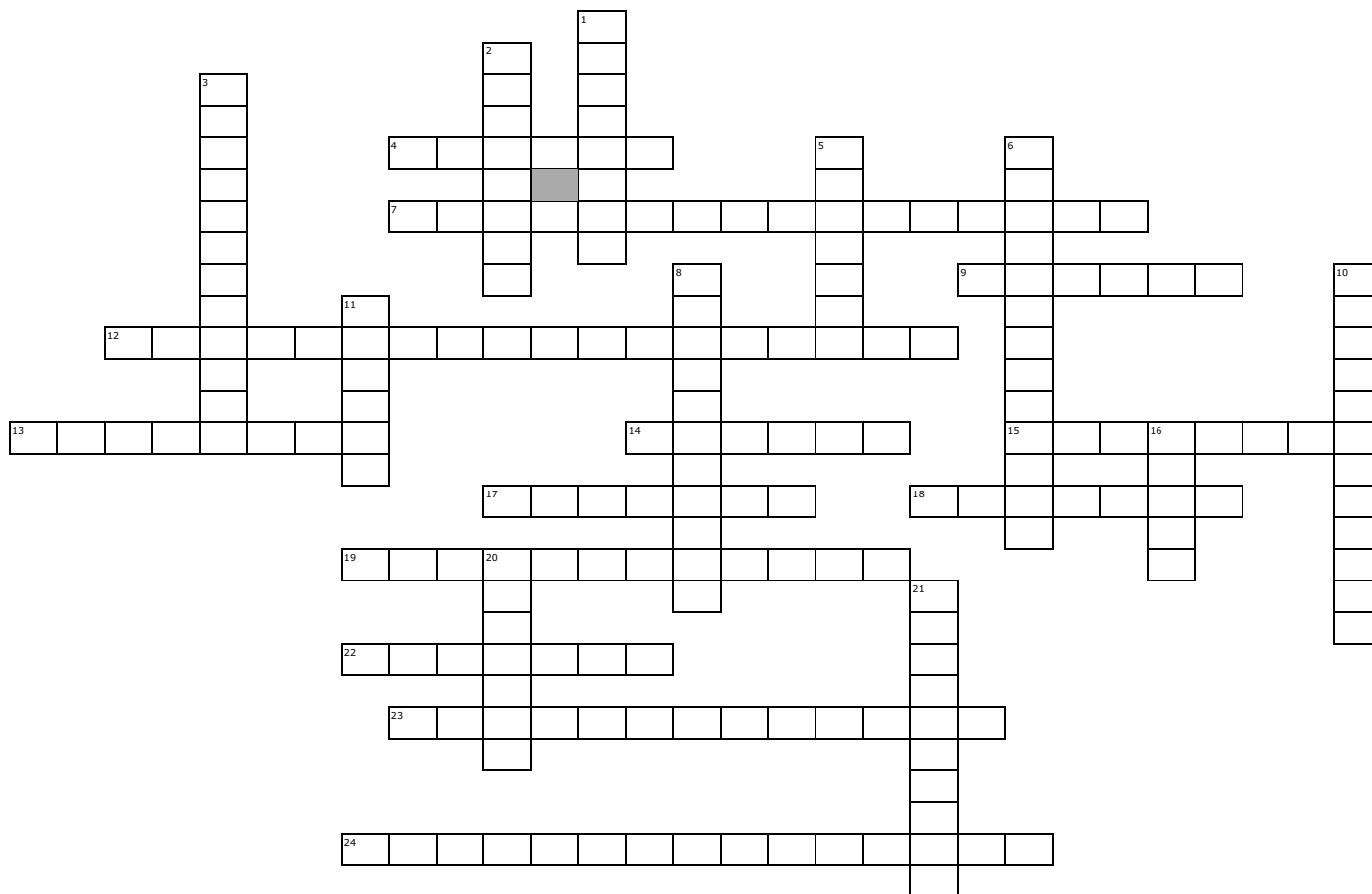


# Intro to Chemistry



## Across

4. Rows on the Periodic Table
7. A type of chemistry that involves carbon based life forms.
9. positively charged particle.
12. Chemistry that relates to non carbon compounds.
13. How quick it is to join up with other elements.
14. Element with 2 protons.
15. A negatively charged particle.
17. The place where Protons and Neutrons are housed
18. A particle that has no charge.
19. Number of Protons in an Element
22. Same number of protons different number of neutrons.
23. Diagram of Elements
24. Creator of the Periodic Table

## Down

1. When two elements bond together.
2. The lightest element known
3. Point at which a liquid is turned into a gas.
5. A substance with just one type of atom.
6. A type of mixture that you can pick apart.
8. A type of mixture that you cannot pick apart. A solution.
10. The point at which a solid turns to a liquid.
11. All things that have mass and volume.
16. Electrons are located around the nucleus in an electron what?
20. When two or more compounds are joined together.
21. Group 18. These elements do not mix well with others.