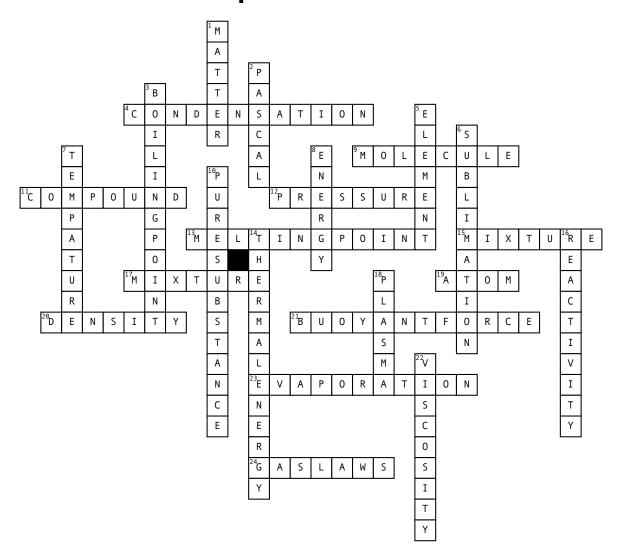
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Chapter 2-3



<u>Across</u>

- **4.** When water vapor in the air beacomes a liquid energy is released from the water to it's surroundings
- **9.** The smallest unit of a substance that behaves like the substance
- ${f 11.}$ A substance made up of atoms of diffrent elements
- ${\bf 12.}$ The amount of force exerted on a given area of surface
- **13.** The tempature at which a substance changes from a solid to liquid
- **15.** A combination of substances that are not chemically combined
- 17. A combination of substances that are not chemically combined

- **19.** The smallest unit of an element that keeps the element's chemical properties
- **20.** A measurement of how much matter is contained in a certain volume of a substance
- 21. Pushes the duck up
- **23.** The change of a substance from a liquid to a gas
- **24.** Describe how the behavior of gases is affected by pressure, volume, and tempature

<u>Down</u>

- Anything that has mass and takes up space
- 2. The SI unit of pressure
- **3.** The tempature at which a liquid changes to a gas

- **5.** A substance that cannot be broken down into simpler substances by chemical means
- **6.** Solids can also change directly into gases in a process called
- 7. A measure of the average kinetic energy of the particles in an object
- $\boldsymbol{8}\text{.}$ The ability to change or move matter, or to do work
- 10. Matter that has a fixed composition and definite properties
- **14.** The total kinetic energy of the particles that make up a substance
- **16.** The capacity of a substance to combine with another substance
- **18.** A state of matter that does not have a definite shape or volume
- 22. A liquid's resistance to flow