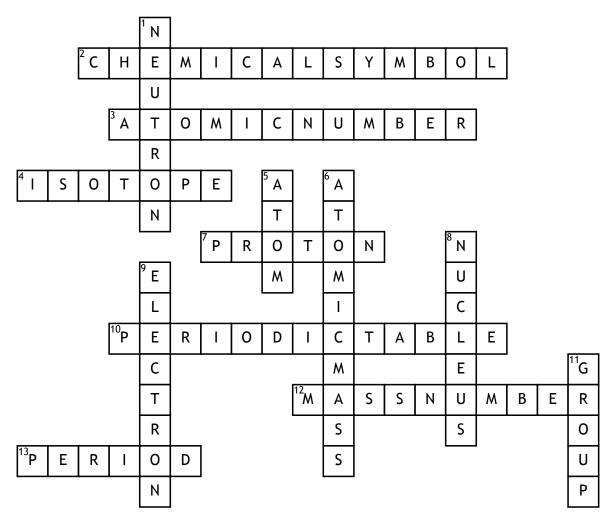
## ATOMS AND THE PERIODIC TABLE



## <u>Across</u>

**2.** A 1- or 2-letter representation of an element

**3.** This represents the number of protons in the nucleus of an atom

4. An atom with the same number of protons and a different number of neutrons from other atoms of the same element

7. A positively charged particle that is part of the nucleus of an atom

**10.** A chart of the elements showing the repeating pattern of their properties

**12.** The sum of the protons and neutrons in the nucleus of an atom

**13.** A series of elements organized in horizontal rows on the periodic table

## <u>Down</u>

**1.** A small particle in the nucleus of the atom with NO electrical charge

**5.** The basic particle from which all elements are made

6. The average mass of all the isotopes of an element

**8.** The central core of the atom

**9.** Negatively charged particles of an atom

**11.** Elements in the same vertical column of the periodic table