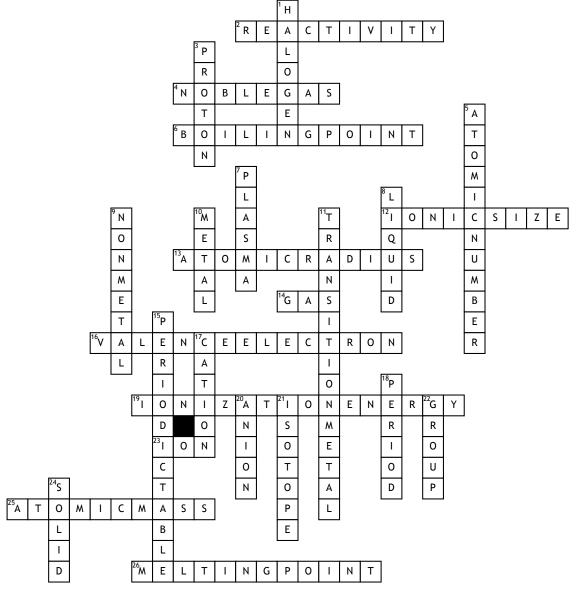
Name:	Date:
-------	-------

## Vocab Cross Word Assignment



## **Across**

- **2.** the tendency of a substance to undergo chemical reaction
- **4.** any of the gaseous elements helium, neon, argon, krypton, xenon, and radon, occupying Group 0 (18) of the periodic table
- **6.** 212 degrees F.
- **12.** When atoms gain or lose electrons
- **13.** a chemical element is a measure of the size of its atoms
- **14.** state of matter with no defined shape or volume
- **16.** an electron that is associated with an atom
- **19.** the energy required to remove an electron from a gaseous atom or ion
- 23. an atom or molecule with a net electric charge due to the loss or gain of one or more electrons

- 25. the mass of a atom
- **26.** the temperature at which a given solid will melt

## Down

- 1. five non-metallic elements found in group 17 of the periodic table
- **3.** a stable subatomic particle occurring in all atomic nuclei
- **5.** the number of protons in the nucleus of an atom
- 7. a hot ionized gas consisting of approximately equal numbers of positively charged ions and negatively charged electrons
- **8.** form of matter between a gas and a solid that has a definite volume, but no definite shape
- **9.** an element that does not possess the properties of metals, and is found in the groups 14, 15 and 16 of the periodic table

- **10.** an element that readily forms positive ions and has metallic bonds
- **11.** any of the set of metallic elements occupying a central block in the periodic table
- **15.** a table of the chemical elements arranged in order of atomic number
- 17. a positively charged ion
- **18.** the name given to a horizontal row of the periodic table
- 20. a negatively charged ion is a
- **21.** One of two or more atoms that have the same atomic number but a different number of neutrons
- **22.** a column of elements in the periodic table
- **24.** state of matter characterized by particles arranged such that their shape and volume are relatively stable