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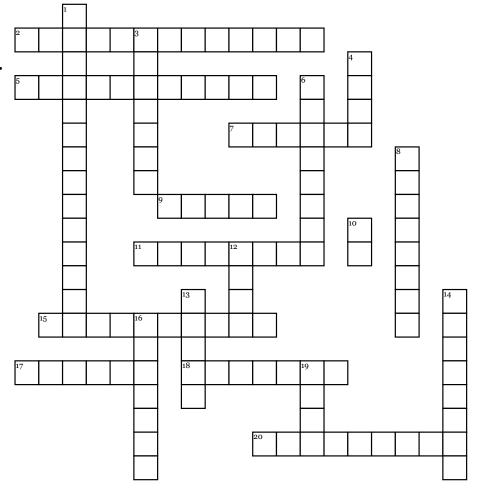
## **Acids And Bases**

## **Across**

- **2.** The process by which a precipitate forms.
- **5.** A common used indicator to determine the pH of acids and/or bases.
- 7. What acids/bases do in substances in solutions to produce either hydrogen or hydroxide(depending on substance).
- **9.** A acid can also be defined as a proton...
- 11. To break down.
- **15.** Chemicals that change colors in the presence of hydrogen and/or hydroxide ions.
- 17. How a base tastes.
- **18.** What a substance is considered if it has a pH of 7.
- **20.** The ion found in a base.

## **Down**

- **1.** The reaction of an acid with a base to produce a salt and water.
- **3.** A scale in which numbers 0-14 are used to determine the pH of a substance.
- **4.** A substance that can accept hydrogen ions because it has an hydroxide ion (-OH).
- **6.** How a base feels.



- **8.** A base can also be defined as a proton...
- **10.** A measure of how acidic or alkaline a substance is.
- 12. How a acid tastes.
- 13. How a acid feels.
- **14.** The ion found in a acid.
- **16.** The word used when acids reacts with active metals.
- **19.** A substance that can donate hydrogen ions (H+).

