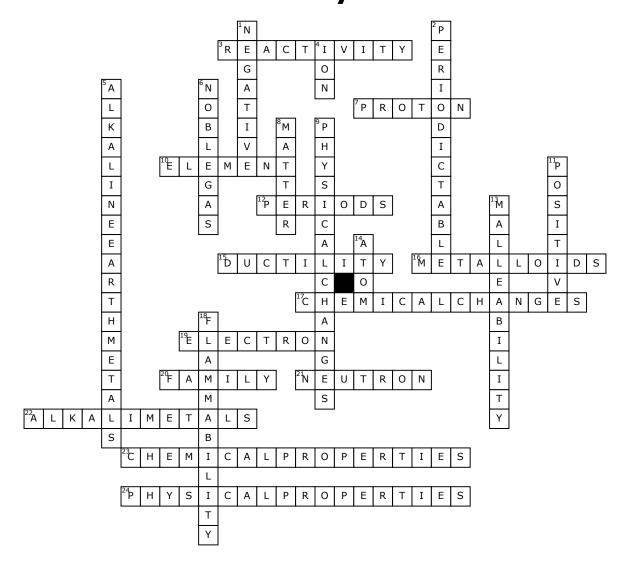
Name:	Date:	Period:

Chemistry Vocab



<u>Across</u>

- **3.** shows the materials's ability to react with another substance
- 7. the positive subatomic particle in the nucleus which determines its atomic number and along with the neutron determines the atomic mass
- **10.** one of a class of substances that cannot be separated into simpler substances by chemical means.
- **12.** the columns of the periodic table; they are grouped by similar chemical and physical properties
- **15.** shows the material's ability to be drawn into a thin wire
- **16.** a group of elements consisting of properties of metals and nonmetals and are located on the diagonal line
- **17.** A usually irreversible change that changes the arrangement of the atoms in the chemical composition that involves a formation of a new substance.

- **19.** the particle in the surrounding cloud of the nucleus having a negative charge and is 1000 times smaller than a proton
- **20.** the rows of the periodic table; they go in an ascending order with the atomic number
- **21.** the particle in the nucleus of the atom which has the biggest mass and no charge which determines the atomic mass with the proton
- **22.** the most reactive metal group located at the leftmost of the the periodic table.
- **23.** a property or characteristic of a substance that is observed during a reaction in which the chemical composition or identity of the substance is changed
- **24.** a property of a substance of matter that can be observed without changing the chemical composition of the substance.

Down

- 1. the charge of an electron
- **2.** a set of elements which are organized on a table according to the atomic number, electron configuration, and physical/chemical properties

- **4.** an atom that is either mostly negatively charged or mostly positively charged
- **5.** any of the group of bivalent metals including barium, radium, strontium, calcium, and, usually, magnesium, the hydroxides of which are alkali's but less soluble than those of the alkali metals.
- **6.** the mostly nonreactive group which are at the rightmost of the periodic table
- 8. anything that has mass and takes up space
- **9.** A usually reversible change in the physical properties of a substance.
- 11. the charge of a proton
- **13.** show the material's ability to be molded into a different shape
- **14.** What is the smallest particle of an element consisting the same chemical properties of the element
- 18. shows the materials's ability to catch fire