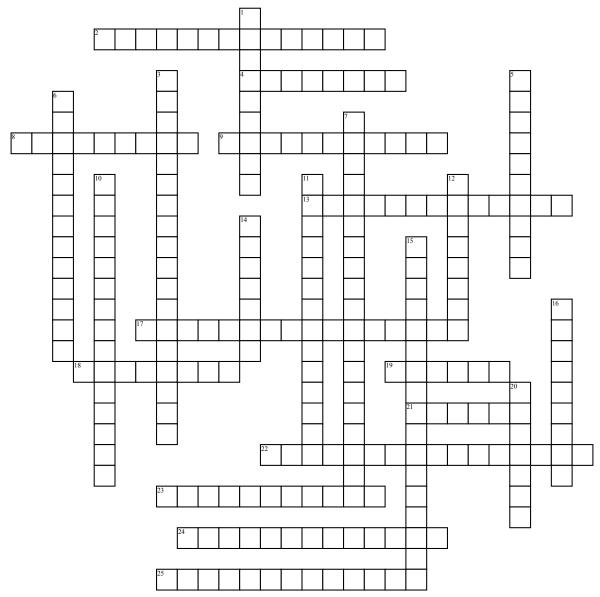
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25 Science Midterm Review Terms



Across

- **2.** Alters the form or appearance of matter, but does not turn any substance in the matter into another substance.
- **4.** A material that increases the rate of a reaction by lowering the activation energy.
- **8.** A substance that enters into a chemical reaction.
- **9.** Are so evenly mixed and difficult to separate the parts of the mixture.
- **13.** Can see different parts and they can be easily be separated out.
- 17. The electrons that are in the highest energy level of an atom and that are involved in chemical bonding.
- **18.** A mixture containing a solvent and at least one solute that has the same properties throughout; a mixture in which one substance is dissolved in another.

- 19. Has definite volume, but no shape of its own, packed almost as close of an arrangement of particles as solids, but it moves more freely.
- 21. Anything that has mass and takes up space.
- **22.** A short, easy way to show a chemical reaction, using symbols.
- **23.** A mixture in which particles can be seen and easily separated by settling or filtration.
- **24.** An arrangement of the elements showing the repeating pattern of their properties.
- **25.** A chemical bond formed when two atoms share electrons.

Down

- 1. Negatively charged particles
- **3.** Releases energy

- **5.** The attraction between oppositely charged ions.
- **6.** 2, 8, 18, and 32
- 7. A change where energy is absorbed.
- **10.** Represents a compound; which shows the elements in the compound and the ratio of atoms.
- **11.** A change in matter that produces one or more new substances.
- 12. Has no electrical charge
- **14.** A substance formed as a result of chemical reaction.
- **15.** A molecule consisting of two atoms.
- **16.** A material that decreases the rate of a reaction.
- 20. Positively charged particles