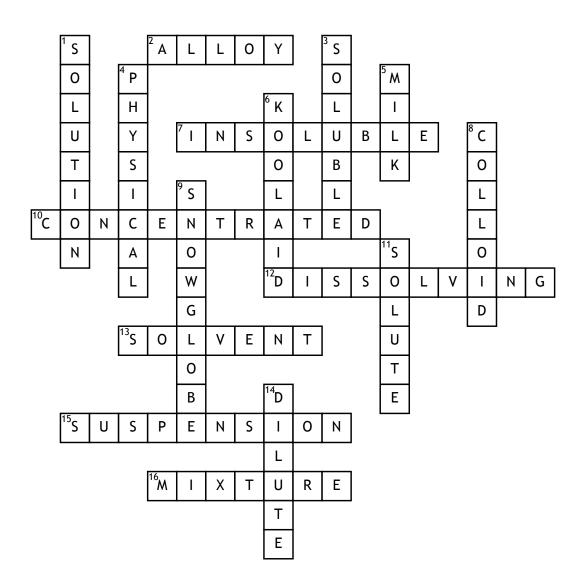
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## **MIXTURES**



## **Across**

- **2.** a solid type of solution in which metal is dissolved into metal
- 7. unable to dissolve
- **10.** a solution with more solute
- **12.** the process in which particles break down and spread evenly throughout a mixture
- **13.** In a solution, the substance that the solute dissolves in

- **15.** a type of mixture in which the particles are too big and heavy to remain mixed
- **16.** a combination of elements, compounds, or both that are not chemically combined (only physically)

## Down

- 1. a type of mixture in which particles are so small and completely dissolved that it looks like one substance
- 3. able to dissolve

- **4.** mixtures can be separated through \_\_\_\_\_ ways such as filters, magnets, and evaporation
- 5. an example of a colloid
- 6. an example of a solution
- **8.** a type of mixture which has properties of both solutions and suspensions
- **9.** an example of a suspension
- 11. In a solution, the substance that is dissolved in a solvent
- 14. a solution with less solute