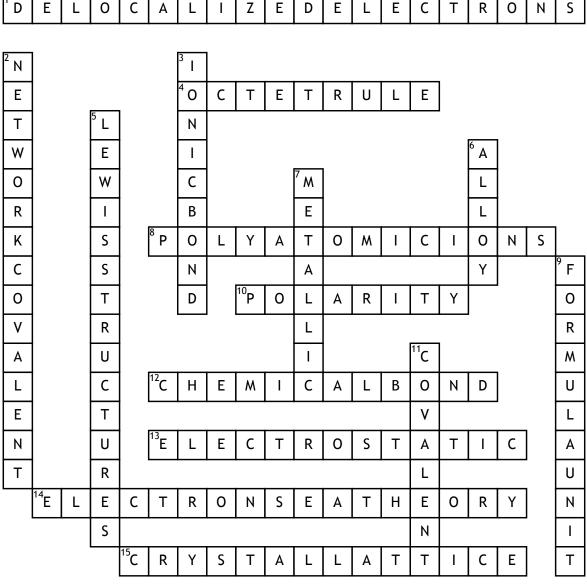
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Chemistry - Chapter 6 Vocabulary



Across

- 1. electrons that can move between several different bonds
- **4.** states that an atom tends to gain, lose, or share electrons until its outer level s and p orbitals are filled with eight electrons
- **8.** groups of atoms that maintain a constant electrical charge while existing as a unit in a wide variety of chemical reactions
- **10.** the tendecy of an object to form two localized regions of opposite character
- **12.** a force of attraction that holds atoms together in compounds that is produced by the transferring or sharing of electrons

- **13.** forces exerted by electrical charges
- **14.** offers an explanation of how metals bond
- **15.** an extensive 3D structure of points or objects that represents the regular, alternating pattern of atoms or ions

Down

- **2.** substances made of atoms covalently bonded into a continuous 3D network
- **3.** the electrostatic attraction between two oppositely charged ions in a solid

- **5.** 2D diagrams that use element symbols and dots to show the bonds between different atoms
- **6.** a mixture of the atoms of a metal with another element where the mixture has metallic properties
- **7.** bond created by a communal sharing of electrons between metal atoms
- **9.** the simplest ration of elements in an ionic compound that describes its composition
- 11. bond created when two atoms held together in an ion or molecule by their attraction for the same pair of shared electrons