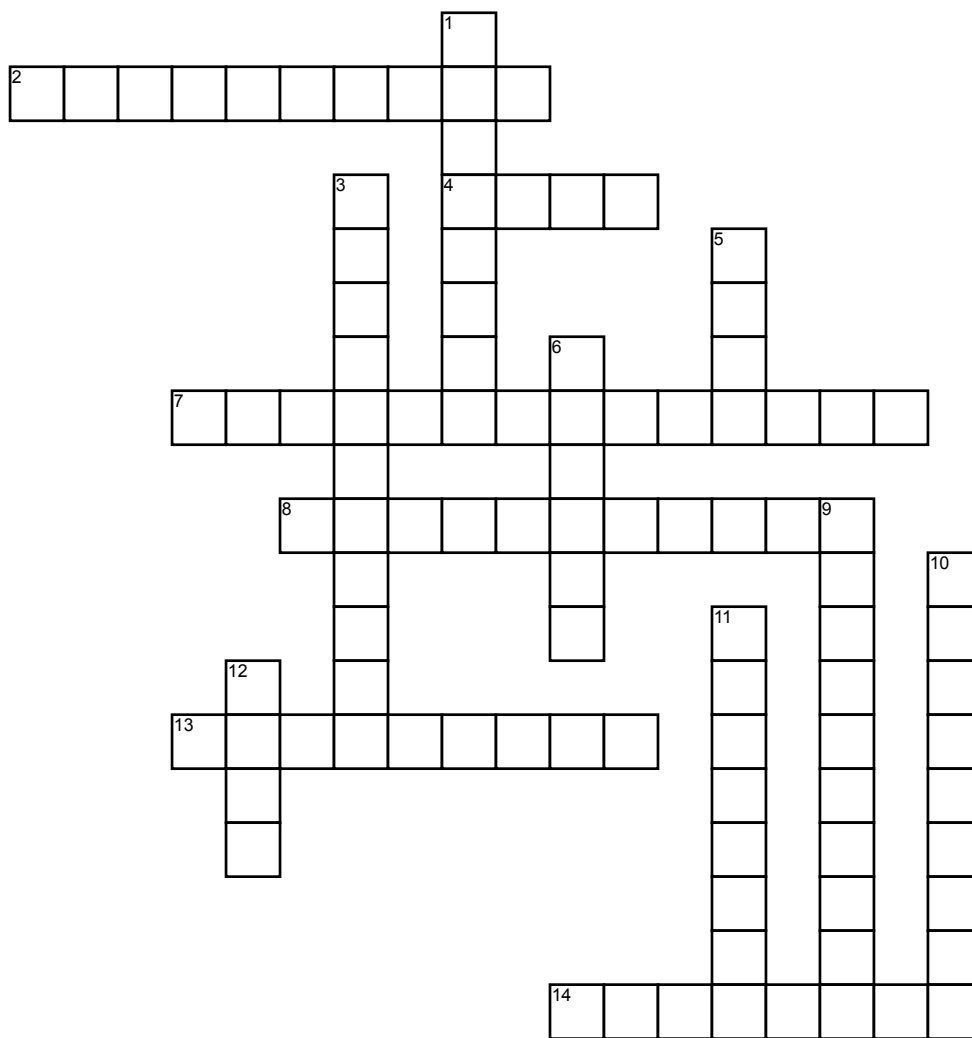


Name: \_\_\_\_\_ Date: \_\_\_\_\_

# Endo and Exothermic Reactions



## Across

2. All chemical reactions require \_\_\_\_\_ energy  
 4. In endothermic reactions the products have \_\_\_\_\_ energy than the reactants  
 7. An example of an endothermic reaction  
 8. In endothermic reactions the \_\_\_\_\_ decreases

13. These bond to make new products

14. Reactants bond to make these

## Down

1. Chemical \_\_\_\_\_ show the reactants and products  
 3. This reaction takes in heat  
 5. An example of an exothermic reaction

6. This is required to make, or break, bonds

9. This reaction puts out heat

10. In exothermic reactions the temperature \_\_\_\_\_

11. Elements that are chemically bonded form these

12. In exothermic reactions the products have \_\_\_\_\_ energy than the reactants

## Word Bank

exothermic	photosynthesis	temperature	more	formulas
products	compound	less	endothermic	increases
energy	fire	reactants	activation	