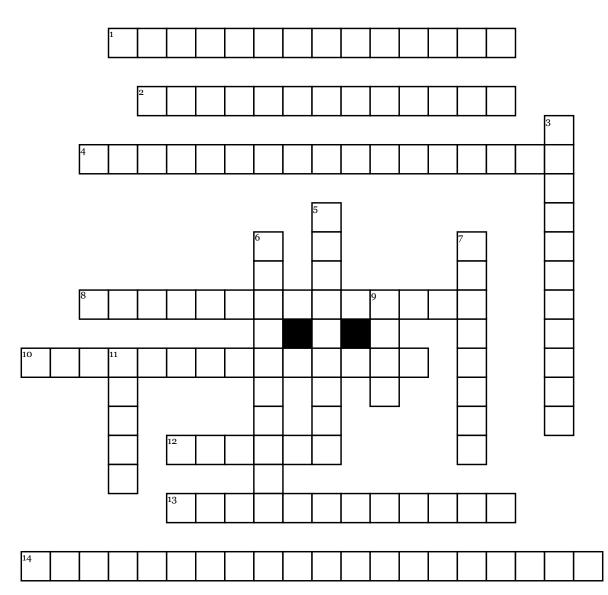
Name:	Date:	Period:
-------	-------	---------

## State of matter



## **Across**

- **1.** A changes affecting the form of a chemical substance but not a chemical composition
- **2.** The sum of the kinetic and potential energy of particles in a materials
- **4.** States that when a force is applied to a confined fluid an increase in pressure is transmitted equally to all parts of the fluid
- **8.** Is any change the at results in the formation of new chemical substances. At the molecular level chemical change involves making or breaking of bonds between atoms.
- **10.** The uneven forces acting on the particles on the surface of a liquid.
- **12.** The ability cause change.
- **13.** Upward force exerted on an object immersed in a fluid.
- **14.** States that the buoyant force on an object is equal to the weight of the fluid displaced by the object

## Down

- **3.** Measure of the average kinetic energy of the individual particles of a substance
- **5.** A liquid's resistance to flow
- **6.** Movement of thermal energy from a substance at a higher temperature to a substance at a lower temperature (movement from high to low.)
- 7. A force exerted on a surface divided by the total area over which the force is exerted.
- **9.** A small particle that makes up most types of matter.
- 11. A pull or push.