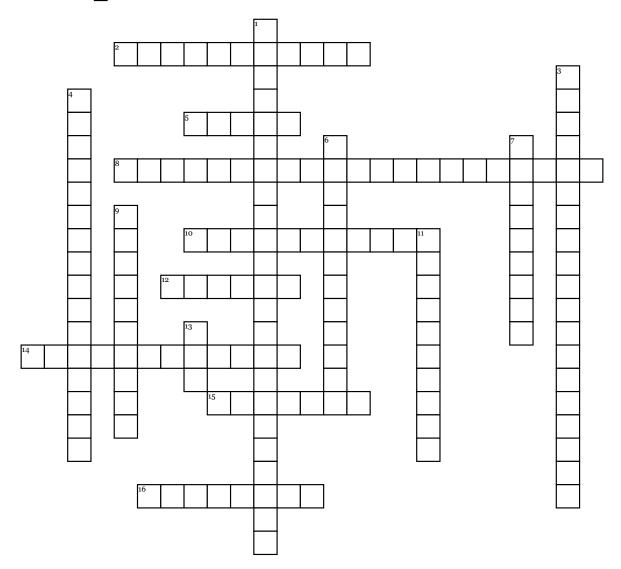
## Chapter 3 Phases of Matter



## **Across**

- **2.** Phase change from a solid to a gas
- **5.** Matter that has a definite shape and volume
- **8.** Fixed in place, packed tightly
- **10.** Heat is absorbed
- **12.** Matter that has no definite shape but a definite volume
- **14.** Phase change from a liquid to a gas

- **15.** Phase change from solid to a liquid
- **16.** Phase change from a liquid to a solid

## **Down**

- 1. Constant motion, closer together, less kinetic energy
- **3.** Constant, random motion
- **4.** A property that can be observed without changing the identity of the substance.

- **6.** Phase change from a gas to a liquid
- 7. The volume of a fixed mass of gas varies inversely with the pressure at constant temperature
- 9. Heat is released
- **11.** The volume of a fixed mass of gas at constant pressure varies directly with the Kelvin temperature scale
- **13.** Matter that has no definite shape or volume