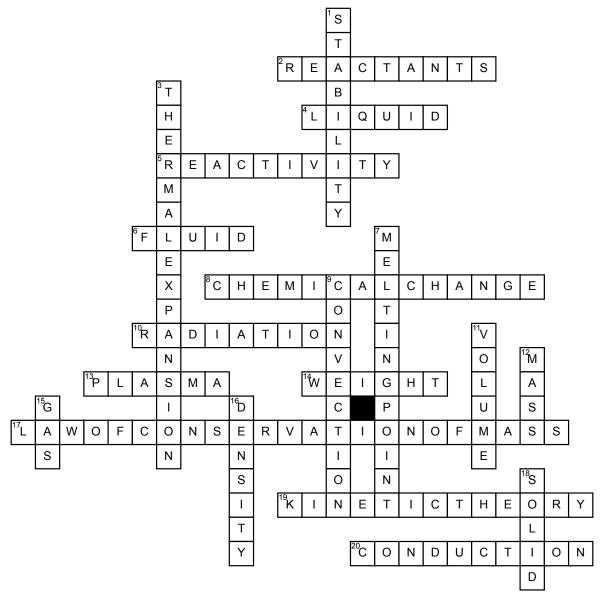
Name:	Date:	Period:

Properties and Changes in Matter



Across

- **2.** Substances that change or react.
- **4.** A definite volume but not a definite shape.
- **5.** Describes how easily a substance reacts with other substances.
- **6.** Flow easily and mix together easily. (gas & liquid)
- **8.** When two or more substances are combined and changed into new substances with new properties.
- **10.** The transfer of thermal energy by electromagnetic waves.
- **13.** A hot mixture of ions and electrons

- **14.** A measure of the force, or pull, on an object due to gravity.
- **17.** Matter can be neither created or destroyed.
- **19.** The idea that particles are made up of matter and are constantly moving.
- **20.** Transfer of thermal energy through particles of matter that are in direct contact.

Down

- **1.** Do not react with other substances.
- **3.** The process that added thermal energy causes most substances to increase size, or expand.

- 7. The temp at which changes between solid in liquid that make it solid melt into a liquid or make a liquid freeze into a solid.
- **9.** The transfer of thermal energy by the motion of fluids (gas & liquid)
- **11.** The amount of space an object takes up.
- **12.** The amount of matter in an object.
- **15.** Has neither a definite shape nor definite volume.
- **16.** The amount of mass per unit volume of a given substance.
- **18.** Definite shape and volume.