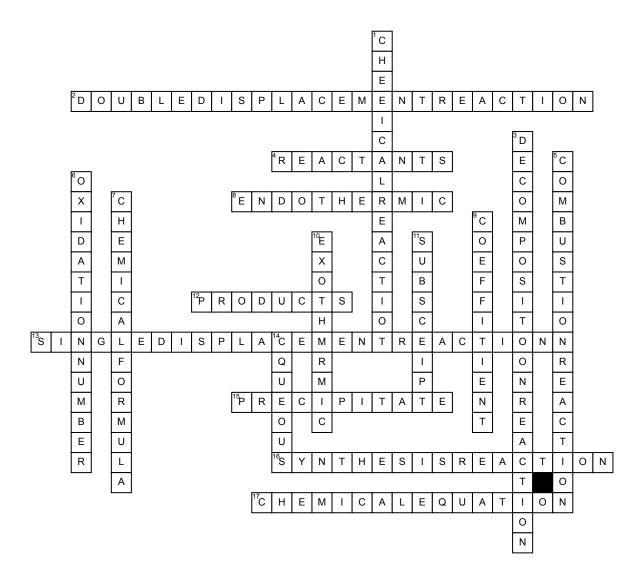
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Chemical Formulas and Chemical Reactions



Across

- 2. a chemical reaction in which the positive ion of one compound replaces the positive ion of another compound
- **4.** the chemicals presented before a reaction
- **8.** a chemical reaction that absorbs energy primarily in the form of heat
- **12.** the new chemicals presented after a reaction
- **13.** a chemical reaction in which one element replaces another element in a compound
- **15.** an insoluble compound that comes out of solution during a double-displacement reaction

- **16.** a chemical reaction in which two or more substances combine to form another substance
- **17.** a way to describe a chemical reaction using chemical formulas and other symbols

Down

- a change in which one or more substances are converted into new substances while energy is released or absorbed
- **3.** a chemical reaction in which one substance breaks down into two or more substances
- a reaction that occurs when a substance reacts with oxygen to release energy in the form of heat and light

- **6.** the amount of valence electrons gained or lost between metals and nonmetals to become stable
- 7. representation of what elements a compound contains and the exact number of the atoms of each elements in a unit of the compound
- **9.** a number that represents the number of units of each chemical taking part in a reaction
- **10.** a reaction in which the energy released is in the form of heat
- 11. means "written below"
- **14.** when water is the solvent for a reaction