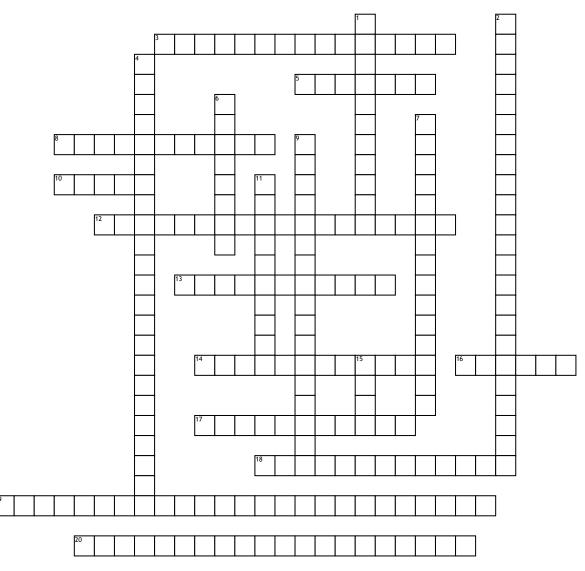
Thermochemistry Vocab.



<u>Across</u>

- 3. stored energy
- 5. unit of heat energy

8. Process of measuring the amount of heat released or absorbed during chemical reactions

10. unit of energy in the SI system

12. enthalpy change that occurs with a system when 1 mole of matter is transformed by a chemical reaction

13. measure of molecular movement

14. The amount of heat needed to raise temp. of one gram of a substance

16. The ability to do work

17. Object process of measuring heat of chemical reactions

18. energy due to motion

19. You can not create or destroy energy

20. one mole of compounds is completely bonded by oxygen gas <u>Down</u>

 when a reaction absorbs heat
form of potential energy that can be absorbed or released during a chemical reaction **4.** amount of heat necessary to boil 1 mole of a substance at its boiling point

6. change where heat is absorbed or released through a chemical reaction

7. The study of energy transfer in the form of heat during reactions and physical changes9. amount of heat to melt 1

mole of a substance 11. when a reaction releases

heat

15. energy that can be transferred due to the molecular movement