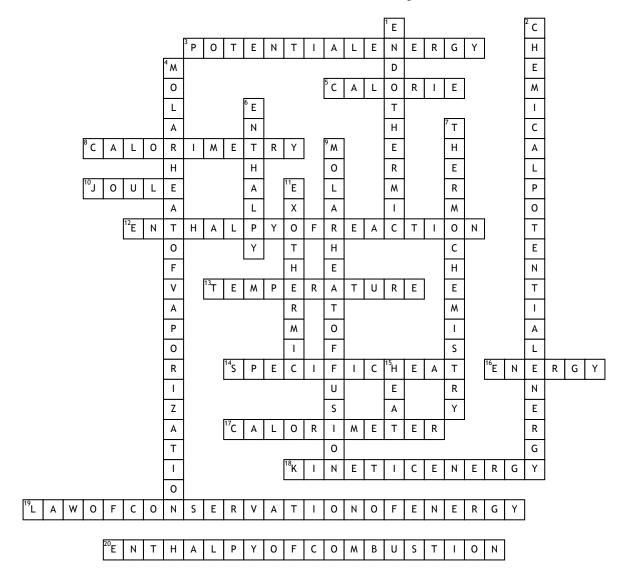
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## Thermochemistry Vocab.



## **Across**

- 3. stored energy
- 5. unit of heat energy
- **8.** Process of measuring the amount of heat released or absorbed during chemical reactions
- **10.** unit of energy in the SI system
- **12.** enthalpy change that occurs with a system when 1 mole of matter is transformed by a chemical reaction
- **13.** measure of molecular movement

- **14.** The amount of heat needed to raise temp. of one gram of a substance
- **16.** The ability to do work
- **17.** Object process of measuring heat of chemical reactions
- 18. energy due to motion
- **19.** You can not create or destroy energy
- **20.** one mole of compounds is completely bonded by oxygen gas
- when a reaction absorbs heat
  form of potential energy that can be absorbed or released during a chemical reaction

- **4.** amount of heat necessary to boil 1 mole of a substance at its boiling point
- **6.** change where heat is absorbed or released through a chemical reaction
- 7. The study of energy transfer in the form of heat during reactions and physical changes
- **9.** amount of heat to melt 1 mole of a substance
- **11.** when a reaction releases heat
- **15.** energy that can be transferred due to the molecular movement