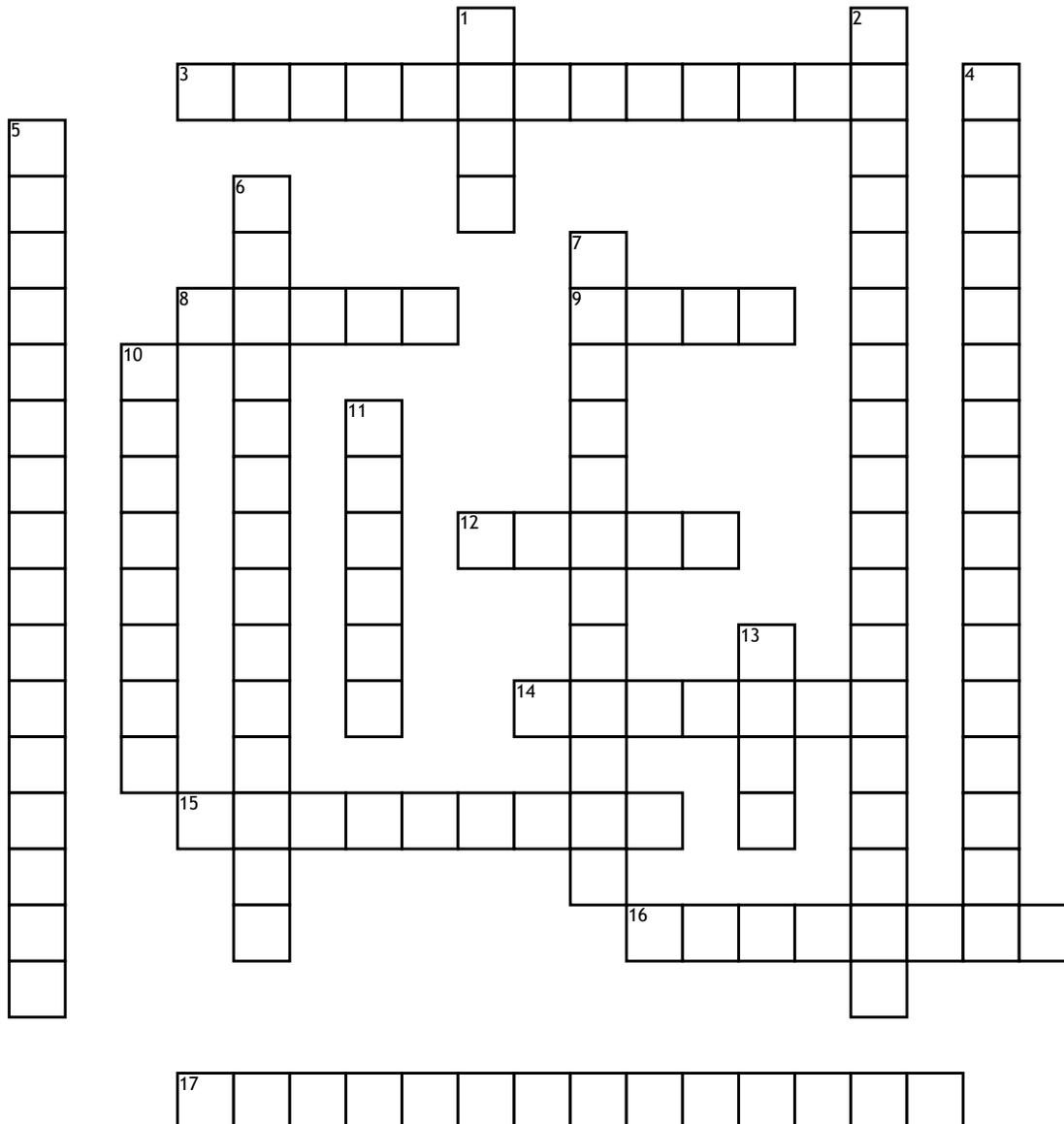


Name: _____

Date: _____

Moles



Across

- 3. 2.50×10^{20} atoms Al
- 8. find the percent composition of Sulfur by using H_2SO_3
- 9. find the percent composition of Hydrogen by using H_2SO_3
- 12. find the percent composition of Sulfur by using $H_2S_2O_8$
- 14. convert to moles: 300.0 grams S
- 15. the mass in grams of one mole of any pure substance

- 16. convert to moles: 5.75×10^{24}

- 17. convert to atoms: 0.230 grams Pb

Down

- 1. find the percent composition of hydrogen by using $H_2S_2O_8$
- 2. the percent by mass of each element in a compound
- 4. a formula that shows the smallest whole-number ratio

- 5. a formula that specifies the actual number of atoms of each element in one molecule of a substance
- 6. convert to atoms: 55.2 g Li
- 7. convert to grams: 42.6 mol Si
- 10. convert to moles: 25.5 g Ag
- 11. convert to grams: 3.57 mol Al
- 13. the SI base unit used to measure the amount of a substance