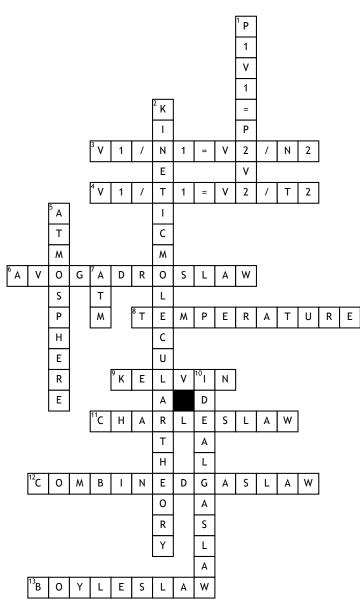
## Gas Laws



## <u>Across</u>

**3.** Formula for Avogadros Law

4. Formula for

Charles Law

6. The law that states equal volumes of gas contain the same number of particles.

**8.** Average kinetic energy of all particles.

**9.** Temperature scale used in gas laws.

**11.** The law that states temperature of a gas varies directly with volume.

12. P1V1/T1=P2V2/T213. The laws which states that pressure of pressure of the pressure of the

a gas varies inversely with volume.

## <u>Down</u>

**1.** Formula for Boyles Law

**2.** The model that describes the behavior of all gases under

certain conditions.

- 5. atm stands for
- 7. Unit for pressure

**10.** PV=nRT