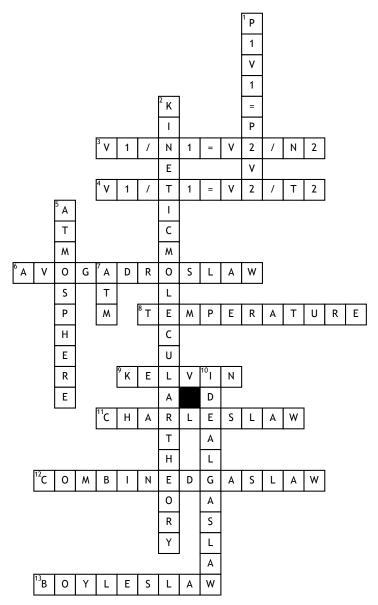
Name:	Date:

## Gas Laws



## **Across**

- **3.** Formula for Avogadros Law
- 4. Formula for Charles Law
- **6.** The law that states equal volumes of gas contain the same number of particles.
- **8.** Average kinetic energy of all particles.

- **9.** Temperature scale used in gas laws.
- 11. The law that states temperature of a gas varies directly with volume.
- **12.** P1V1/T1=P2V2/T2
- **13.** The laws which states that pressure of a gas varies inversely with volume.

## <u>Down</u>

- 1. Formula for Boyles Law
- **2.** The model that describes the behavior of all gases under certain conditions.
- 5. atm stands for
- 7. Unit for pressure
- 10. PV=nRT