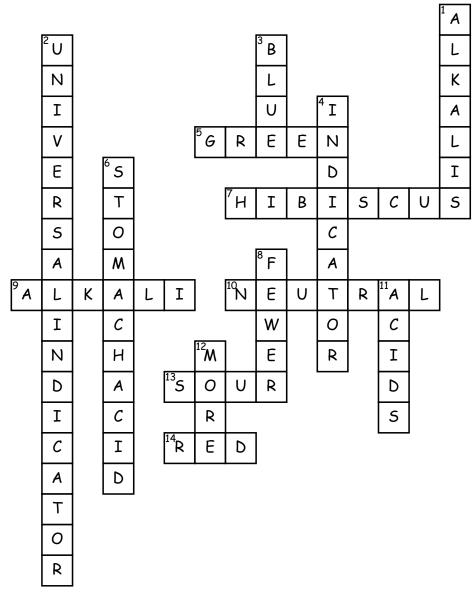
## Acids and Alkali



## <u>Across</u>

**5**. The colour of universal indicator at pH 7

7. This flower goes pink/ red in acid and dark green in alkali solution

**9**. Washing-up liquid is an example of an

**10**. Water is \_\_\_\_\_ because it has a pH of 7

- 13. Acids taste \_\_\_\_
- 14. Blue litmus paper turns \_\_\_\_\_\_ when added to acids

## <u>Down</u>

- 1. The have a pH above 7
- 2. we can use a

HOW acidic or alkaline a solution is

**3**. Red litmus paper turns \_\_\_\_\_ when added to alkalis

4. We can use an

\_\_\_\_\_\_ to find out if something is acid or alkaline

6. Which is more acidic, milk or stomach acid?
8. A concentrated solution has \_\_\_\_\_\_ acid particles per litre than a dilute solution

**11**. The have a pH of less than 7

12. A concentrated solution of acid is \_\_\_\_\_ corrosive than a dilute solution