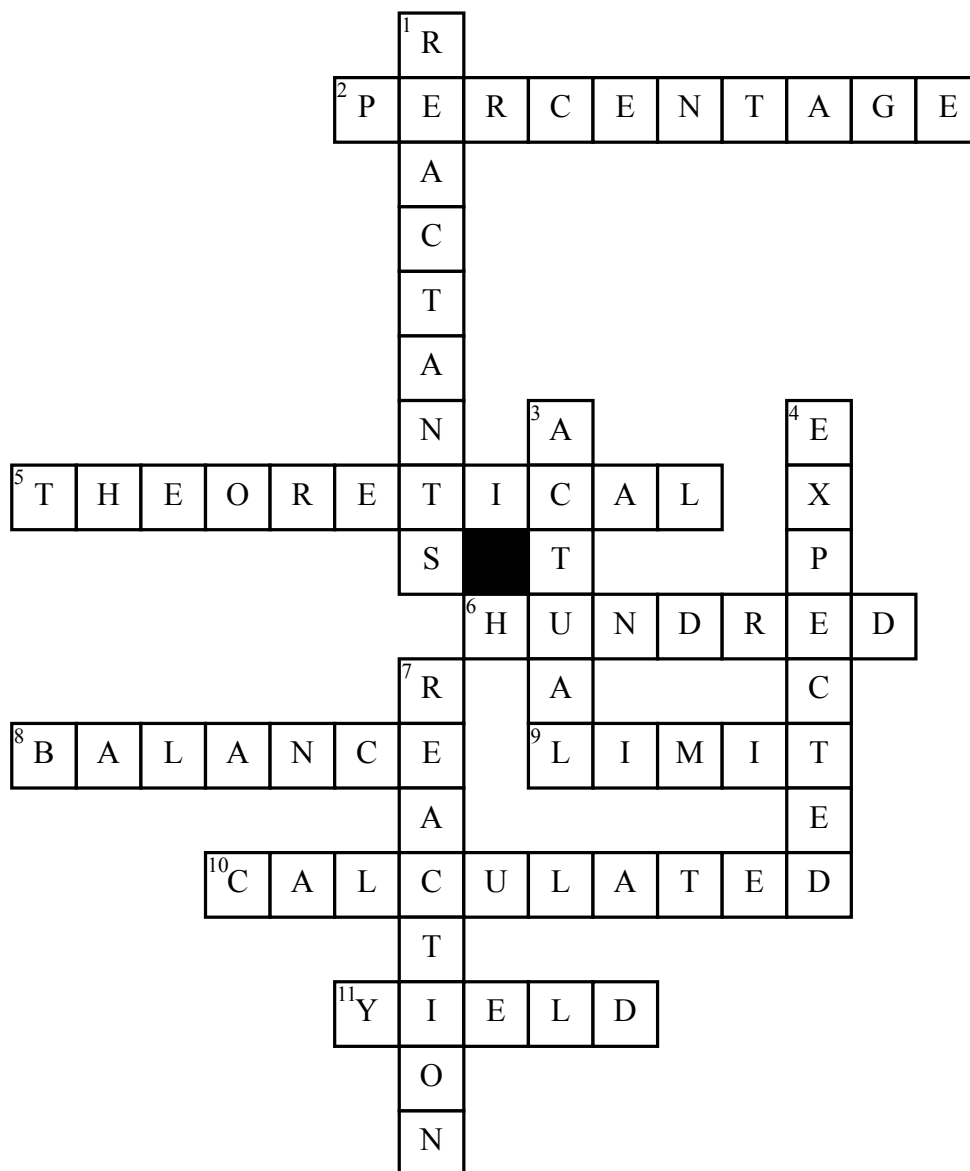


Name: _____ Date: _____ Period: _____

Theoretical and Percent Yield



Across

2. 5th step you find the _____ yield
5. 3rd step you find the _____ yield
6. The actual yield will never be a _____ percent due to limitations
8. 1st step you _____ the equation
9. 2nd step you _____ reagent
10. The theoretical yield for a reaction is _____ based on the limiting reagent
11. 4th step you find the actual _____

Down

1. the _____ may not always be present in the proportions written in the balanced equation
3. _____ yield/ theoretical yield
4. the _____ amount of product is the theoretical yield
7. Percent yield measures how effect the _____ is under certain conditions