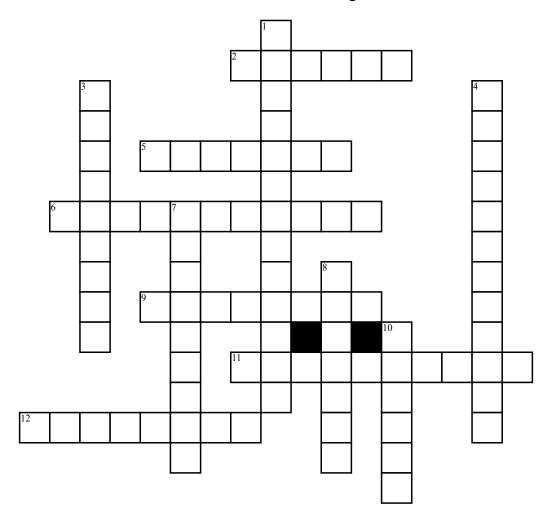
Name: _____ Date: ____ Period: _____

solubility



Across

- 2. substance that dissolves
- **5.** substance that does the dissolving
- **6.** the solid that comes out of a solution
- **9.** a mixture of substances where one is dissolved into the other
- **11.** the concentration of a saturated solution
- **12.** when particles of a solute interact with and are completely surrounded by particles of a solvent

Down

- 1. the mass of solute dissolved per unit volume of solvent
- **3.** describes a solution that will hold no more solute
- **4.** containing a large amount of of solute compared to a solvent
- **7.** does NOT have ability to dissolve
- 8. has ability to dissolve
- **10.** containing a small amount of solutecompared to a solvent

Word Bank

dissolve solvent soluble concentration concentrated solute precipitate saturated insoluble dilute solution solubility