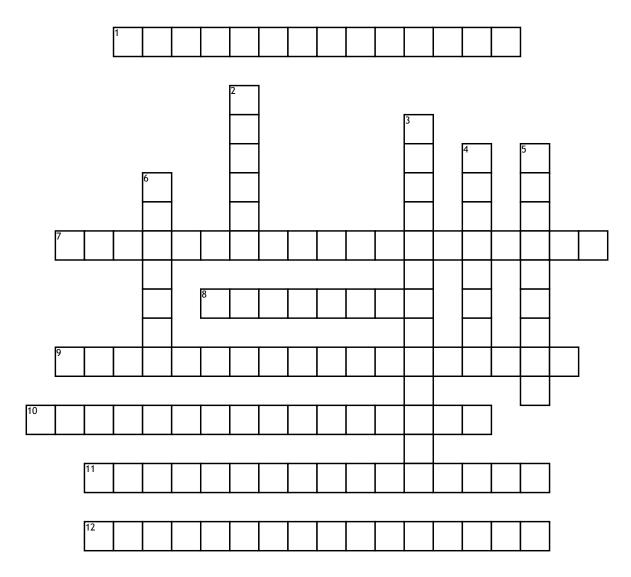
Name:	Date:
-------	-------

Chemical Reactions



Across

- 1. measure of how fast a chemical reaction occurs
- **7.** chemical reaction in which heat energy is absorbed
- **8.** substance that speeds up a chemical reaction but is not used up itself or permanently changed
- 9. chemical reaction in which heat energy is released 10. process that produces chemcial change, resulting in new substances that have properties different from those of the original substances

- 11. minimum amount of energy needed to start a chemical reaction
- 12. shorthand form of writing what reactants are used and what products are formed in a chemical reaction; sometimes shows whether energy is produced or absorbed

Down

2. a type of protein that regulates chemical reactions in cells without being changed or used up itself

- **3.** describes how much solute is present in a solution compared to the amount of solvent
- **4.** substance that exists before a chemical reaction begins
- **5.** substance that slows down a chemical reaction, making the formation of a certain amount of product take longer
- **6.** substance that froms as a result of a chemical reaction