

Name: _____ Date: _____

Noble Gases and Halogens Activity

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| 1. Noble gases are unreactive because H | A. Odorless and colorless |
| 2. Halogen that is solid at room temperature G | B. Yellow-Green Gas |
| 3. Halogens D | C. Radon |
| 4. Having multiple names E | D. Need 1electron to be stable |
| 5. Krypton M | E. Noble Gases |
| 6. Naturally occurring as diatomics K | F. Xenon |
| 7. Chlorine B | G. Bromine |
| 8. Group (7) Elements I | H. 8 electrons in valence shell |
| 9. Group 0 A | I. Kill bacteria,bleaching agent |
| 10. Atomic number 54 F | J. Atomic Mass 35.4 amu |
| 11. Chlorine J | K. Halogens |
| 12. Highly radioactive C | L. 2,7 |
| 13. Flourine electron Arrangement in the shells L | M. Used in Fluorescent lamps |