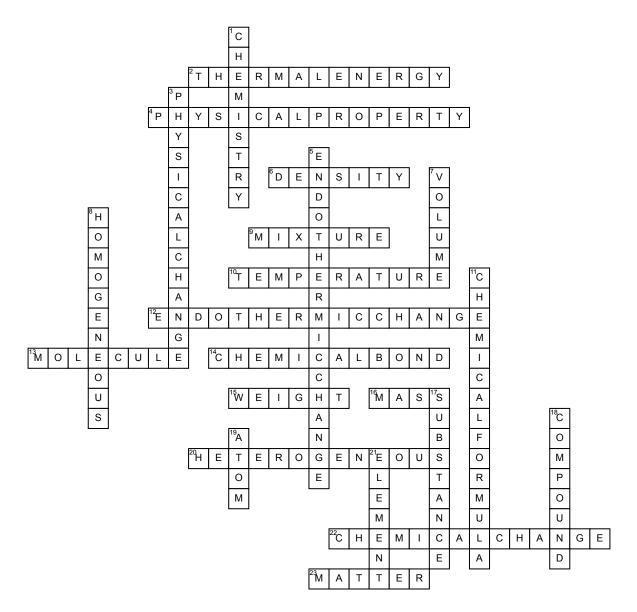
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Substances and Mixtures



Across

- **2.** The total energy of the motion of all the particles in an object
- **4.** A characteristic of a substance that can be observed without changing it into another substance
- **6.** A measure of the mass of a material in a given volume
- **9.** Made of two or more substances that are together in the same place, but their atoms are not chemically bonded
- **10.** A measure of how hot or cold something is
- **12.** A change in which energy is absorbed
- **13.** A group of two or more atoms held together by chemical bonds

- **14.** When atoms combine, they form this, which is a force of attraction between two atoms
- **15.** A measure of the force of gravity on you
- **16.** The amount of matter in an object
- **20.** You can usually see the different parts and they can be easily be separated
- **22.** A change in matter that produces one or more new substances
- **23.** Anything that has mass and takes up space

Down

- **1.** The study of matter and how matter changes
- **3.** Alters the form appearance of matter but does not turn any substances in the matter into a different substance

- 5. Releases energy
- **7.** The amount of space that matters occupies
- **8.** Are evenly mixed that you can't see the different parts
- **11.** Which shows the elements in the compound and the ratio of atoms
- **17.** Å single kind of matter that is pure, meaning it always has a specific makeup, or composition
- **18.** A substance made of two or more elements that are chemically combined in a set ratio
- **19.** The basic particle from which all elements are made
- **21.** A substance that cannot be broken down into any other substances by chemical or physical means