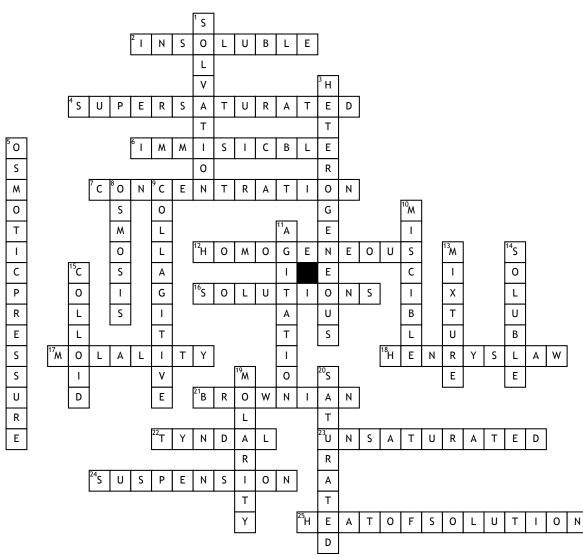
## Solubility Crossword



## <u>Across</u>

**2.** substances that dont dissolve in a solvent

**4.** over max solute at given temprature

6. two substances that can be mixed and seperated

7. how much solute is in the solvent

**12.** mixture with constant concentration throughout

**16.** another name for homogeneous

mixtures

**17.** the number of moles of solute dissolved per kilogram of solution

**18.** law that states at a given temprature, solubility of a gas is proportional with pressure

**21.** type of motion that is the irratic movment of colloids

**22.** effect that is when scattered colloid particles produce light

23. less than max solute

**24.** mixture containing particles that settle out

**25.** the overall energy change in the solution process

<u>Down</u>

1. surrounding solute particles with solvent particles

3. mixture that does not blend smoothly

**5.** pressure added by the newly introduced solvent in osmosis

**8.** diffusion of a solvent through a semi-permiable membrane

**9.** physical properties of a solution that has to do with the number of particles

**10.** two liquids that are soluble in each other

**11.** stirring or mixing

**13.** combination of two or more pure substances

**14.** a substance that can be dissolved is

**15.** Heterogeneous mixture of intermediate sixed particles

19. the number of moles of solute dissolved per liter of solution20. max solute